

THE FACILE SYNTHESIS OF 2'-AZIDO-2',3'-DIDEOXYADENOSINE.

PREPARATIVE APPLICATIONS OF THE DEOXYGENATIVE [1,2]-HYDRIDE SHIFT AND β -ELIMINATION REACTIONS OF SULFONATES WITH $Mg(OMe)_2-NaBH_4$

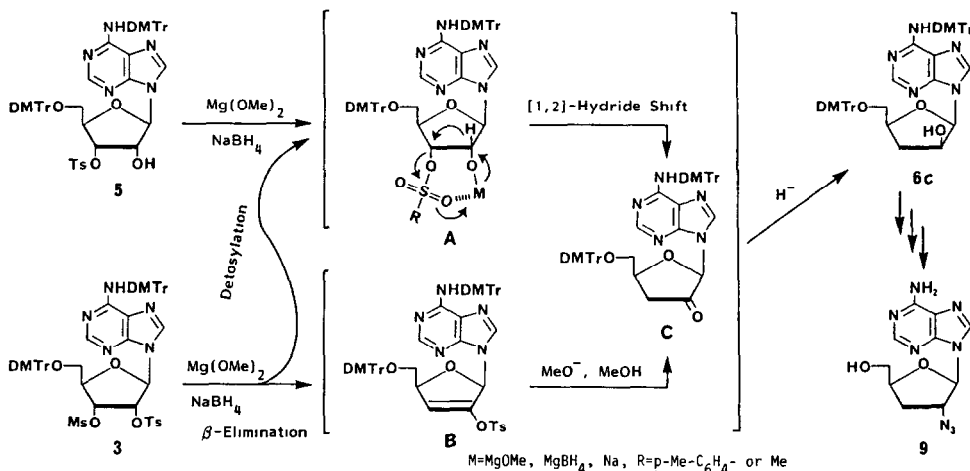
Masajiro Kawana* and Hiroyoshi Kuzuhara

RIKEN (The Institute of Physical and Chemical Research), Wako-shi, Saitama 351-01, Japan

Abstract: The title compound was easily prepared in 37 or 51% overall yield from 2'- or 3'-O-tosyladenosine, respectively, in 5 - 6 steps without isolation of any intermediates. A key intermediate was synthesized by the title reactions in a one-pot procedure.

Recently much attention has been given to the synthesis and biological evaluation of 2',3'-dideoxysugar nucleosides and their azido analogues, because some of these have been shown to possess significant inhibitory activity against AIDS (Acquired Immune Deficiency Syndrome)-associated virus.¹⁾ We wish to report an efficient and practical method for synthesizing the hitherto unknown title compound **9** via a versatile intermediate, N^6, O^5' -bis(4,4'-dimethoxytrityl)-9-(3-deoxy- β -D-threo-pentofuranosyl)adenine **6c**, which would be useful for the preparation of other deoxy- or dideoxysugar nucleosides. An N^6, O^5' -bis-monomethoxytritylated derivative **6b** has been known, but this required an 8-step synthesis from adenosine and isolation of the final product from a diastereomeric by-product.²⁾

A new route to **6c** involves a 2 or 3-step synthesis in a one-pot procedure. We have found that combined reagents, magnesium methoxide-sodium borohydride [$Mg(OMe)_2-NaBH_4$], were extremely effective for the conversion of N^6, O^5' -bis(4,4'-dimethoxytrityl)-3'-O-tosyladenosine **5** to **6c** through a 2'-keto nucleoside **C**,³⁾ which was formed by a [1,2]-hydride shift as formulated in **A**.^{4,5)} The same reagent system was found to be very efficient for preparing **6c** from N^6, O^5' -bis(4,4'-dimethoxytrityl)-3'-O-mesyl-2'-O-tosyladenosine **3** via an enol sulfonate **B**,⁶⁾ which was formed by the β -elimination of **3**.



Starting materials, 2'-O-tosyladenosine 1⁷⁾ and its isomeric 3'-O-tosylate, were readily obtained from adenosine in 2 steps. The latter was isolated as its *p*-toluenesulfonate 4⁸⁾ (about 10% yield) from a mother liquor for the crystallization of crude 1. All of the reactions for synthesizing the azido compound 9 proceeded very smoothly; therefore, purification in each step of the reactions was unnecessary.

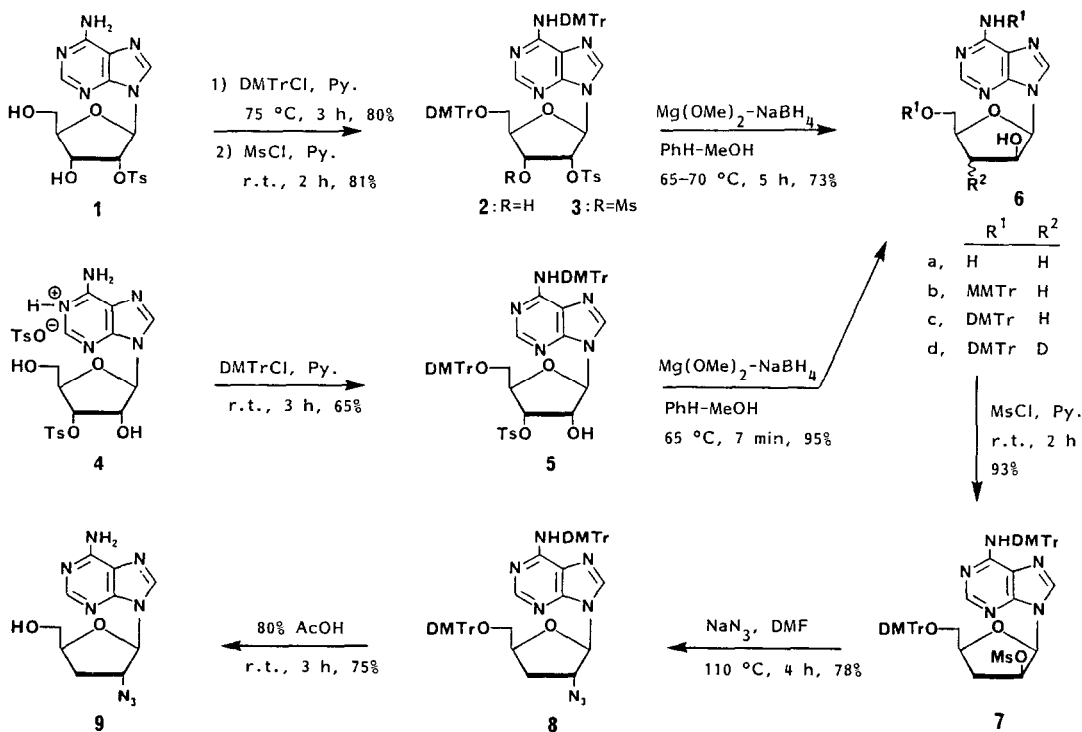
A typical procedure is as follows. To a suspension of 1 (1.68 g, 4 mmol) in dry pyridine (20 ml) was added 4,4'-dimethoxytrityl chloride (DMTrCl; 2.98 g, 8.8 mmol), and the mixture was heated at 75 °C for 2 h, and then cooled to room temperature. Without isolation of the resulting bis-dimethoxytritylated derivative 2, mesyl chloride (0.62 ml, 8 mmol) was added, and the mixture was stirred at room temperature for 3 h. After the usual work-up, the pyridine was removed by co-evaporation with toluene to give crude 3, which was dissolved in a mixture of benzene (35 ml) and methanol (35 ml). To this solution there were added Mg(OMe)₂ (2.75 g, 32 mmol) and NaBH₄ (760 mg, 20 mmol), and the mixture was gently heated at 65–70 °C (bath temperature) under an atmosphere of dry nitrogen for 5 h; hydrogen gas generated during the heating. After cooling, the mixture was extracted with diethyl ether containing a small amount of chloroform, and the organic layer was washed successively with aqueous ammonium chloride and water. The usual work-up gave crude 6c, which was, without purification, treated with mesyl chloride (0.82 ml, 10.6 mmol) in dry pyridine (20 ml) at room temperature for 2 h to give crude N⁶,O^{5'}-bis(4,4'-dimethoxytrityl)-9-(3-deoxy-2'-O-mesyl-β-D-threo-pentofuranosyl)adenine 7. A mixture of the total crude 7 and sodium azide (1.3 g, 20 mmol) in dry N,N-dimethylformamide (DMF, 20 ml) was stirred at 110–115 °C (bath temperature) for 4 h. After cooling, the mixture was diluted with diethyl ether and a small amount of chloroform, and the DMF was removed by extraction with water. The organic solution was dried over magnesium sulfate, and concentrated to give crude N⁶,O^{5'}-bis(4,4'-dimethoxytrityl)-9-[(2R)-2-azido-2,3-dideoxy-β-D-glycero-pentofuranosyl]adenine 8, which was subjected to the final deprotection reaction. A mixture of the total crude 8 in 80% acetic acid (60 ml) was stirred at room temperature for 3 h, and the acetic acid was co-evaporated with toluene-ethanol to afford a solid, which was triturated with dichloromethane (20 ml). The undissolved materials were collected by filtration and dissolved in refluxing ethanol. On cooling, 9 was spontaneously crystallized: 413 mg (37% overall yield from 1).

Alternative way to get 9 was easier than the previous one. A mixture of 4 (2.39 g, 4 mmol) and DMTrCl (2.85 g, 8.4 mmol) in dry pyridine (20 ml) was stirred at room temperature for 3 h. After the usual work-up and removal of the pyridine, the resulting crude N⁶,O^{5'}-bis-(4,4'-dimethoxytrityl)-3'-O-tosyladenosine 5 was treated with Mg(OMe)₂ (2.75 g, 32 mmol) and NaBH₄ (380 mg, 10 mmol) in a mixture of benzene (35 ml) and methanol (35 ml) at 65 °C for 10 min under an atmosphere of dry nitrogen to yield crude 6c, which was converted to 9 (560 mg, 51% overall yield from 4) in a manner similar to that for the preparation starting from 1.

Each reaction was carried out independently by the use of the pure corresponding starting materials, and the analytically pure products, 2, 3, and 5 – 8 were fully characterized.⁸⁾ In the deoxygenative reductions, no epimer of 6c was detected by means of ¹H NMR spectroscopy. The configuration of a 2'-up OH group in 6c was ascertained by the conversion of 6c into the known 9-(3-deoxy-β-D-threo-pentofuranosyl)adenine 6a.²⁾ The R configuration at the 2'-position of 8 was deduced on the basis of the stereochemical course of a S_N2 reaction at the C-2' of 7.

The combined reagents, $Mg(OMe)_2-NaBH_4$, in methanol are considered to produce sodium methoxide in situ according to the HSAB principle.⁹⁾ It seems likely that the strongly basic methoxide ion attacked not only the H-2' of **3** for the β -elimination, but also its tosyl sulfur to form **A** (R=Me). It has been known that 2'-O-acyl^{10a)}, benzyl^{10b)}, and tosyl^{10c)} groups in nucleosides were removed under basic conditions. The incorporation of a deuterium atom at the C-3' of **6c** in the path **B** \rightarrow **C** was examined by the use of methanol- d_1 instead of methanol. The 1H NMR spectrum of the product from **3** showed that the content of the deuterium at this position (**6d**) was 37.5%. This suggested that about 75% of the reaction proceeded through **B**, while most of the remainder underwent via **A**.

These procedures are presently being applied to the modification of other ribonucleosides to the corresponding 3'-deoxy or 2',3'-dideoxy analogues.



DMTr=4,4'-dimethoxytrityl; MMTr=4-monomethoxytrityl; Ms=mesyl; Ts=tosyl.

References and Notes

- 1) R. K. Robins, *Chem. Eng. News*, **1986**, 28; E. D. Clercq, *J. Med. Chem.*, **29**, 1561 (1986); T.-S. Lin, M. S. Chen, C. McLaren, Y.-S. Gao, I. Chazzouli, and W. H. Prusoff, *ibid.*, **30**, 440 (1987); H. Mitsuya and S. Broder, *Nature*, **325**, 773 (1987).
- 2) A. Nyilas and J. Chattopadhyaya, *Synthesis*, **1986**, 196.
- 3) M. Kawana, K. Takeuchi, T. Ohba, and H. Kuzuhara, *Nucleic Acids Res. Sym. Ser. No. 17*, 37 (1986).

- 4) M. Kawana and S. Emoto, Tetrahedron Lett., 3395 (1975); idem., Chem. Lett., **1977**, 597; idem., Bull. Chem. Soc. Jpn., **53**, 222 (1980).
- 5) F. Hansske and M. J. Robins, J. Am. Chem. Soc., **105**, 6736 (1983).
- 6) T. Sasaki, K. Minamoto, and S. Tanizawa, J. Org. Chem., **38**, 2896 (1973).
- 7) D. Wagner, J. P. H. Verheyden, and J. G. Moffatt, J. Org. Chem., **39**, 24 (1974); S. Uesugi, T. Kaneyasu, J. Matsugi, and M. Ikehara, Nucleosides and Nucleotides, **2**, 373 (1983).
- 8) Satisfactory elemental analyses were obtained for all new compounds, and their spectral data are as follows [mp; specific rotations (CHCl₃); UV, $\frac{\text{MeOH}}{\text{max}} (\epsilon)$; ¹H NMR (400 MHz, CDCl₃); IR, $\nu_{\text{max}}^{\text{KBr}}$]. **2**: amorphous; $[\alpha]_{\text{D}}^{24} -39.2^\circ$ (c 0.5); 274 nm (26100); δ 2.29 (3H, s, O-CH₃), 3.31 (1H, dd, H-5'), 3.44 (1H, dd, H-5''), 3.77 (12H, s, O-CH₃), 4.19 (1H, m, H-4'), 4.65 (1H, m, H-3'), 5.73 (1H, t, H-2'), 6.07 (1H, d, $J=6.1$ Hz, H-1'), 7.73 (1H, s, H-2 or 8), 7.81 (1H, s, H-8 or 2). **3**: amorphous; $[\alpha]_{\text{D}}^{20} -34.1^\circ$ (c 0.6); 273 nm (27900); δ 2.28 (3H, s, Ts-CH₃), 3.11 (3H, s, Ms-CH₃), 3.45 (1H, dd, H-5'), 3.54 (1H, dd, H-5''), 3.78 (12H, s, O-CH₃), 4.45 (1H, q, H-4'), 5.44 (1H, dd, H-3'), 5.89 (1H, dd, H-2'), 6.08 (1H, d, $J=6.6$ Hz, H-1'), 7.69 (1H, s, H-2 or 8), 7.78 (1H, s, H-8 or 2). **4**: 195—196 °C (dec.); $[\alpha]_{\text{D}}^{27} -23.0^\circ$ (c 0.94, DMSO); 259 nm (16200); δ (DMSO-d₆ - D₂O) 2.30 (3H, s, CH₃), 2.43 (3H, s, CH₃), 3.42 (1H, dd, H-5'), 3.58 (1H, dd, H-5''), 4.14 (1H, m, H-4'), 4.82 (1H, dd, H-2'), 5.03 (1H, dd, H-3'), 5.96 (1H, d, $J=7.0$ Hz, H-1'), 7.15 (2H, d,), 7.51 (4H, m,), 7.88 (2H, m,), 8.46 (1H, s, H-2), 8.64 (1H, s, H-8). **5**: 130 (sintered)—142 °C; $[\alpha]_{\text{D}}^{22} +4.8^\circ$ (c 0.57); 274 nm (24200); δ 2.40 (3H, s, O-CH₃), 3.07 (1H, dd, H-5'), 3.39 (1H, dd, H-5''), 3.76 and 3.77 (12H, each s, O-CH₃), 4.42 (1H, m, H-4'), 5.02 (1H, q, H-2'), 5.08 (1H, dd, H-3'), 5.15 (1H, d, OH), 5.88 (1H, d, $J=6.1$ Hz, H-1'), 7.97 (2H, s, H-2 and 8). **6c**: amorphous; $[\alpha]_{\text{D}}^{22} +9.2^\circ$ (c 0.9); 274 nm (30000); δ 2.15 (1H, m, H-3'), 2.47 (1H, m, H-3''), 3.19 (1H, dd, H-5'), 3.56 (1H, dd, H-5''), 3.78 (12H, s, O-CH₃), 4.34 (1H, m, H-4'), 4.53 (2H, br s, H-2' and OH), 6.06 (1H, d, $J=2.9$ Hz, H-1'), 8.02 (1H, s, H-2), 8.24 (1H, s, H-8). **7**: amorphous; $[\alpha]_{\text{D}}^{22} +8.2^\circ$ (c 0.9); 274 nm (29400); δ 2.47 (1H, m, H-3'), 2.62 (1H, m, H-3''), 2.52 (3H, s, O-CH₃), 3.38 (1H, dd, H-5'), 3.44 (1H, dd, H-5''), 3.78 and 3.79 (12H, each s, O-CH₃), 4.34 (1H, m, H-4'), 5.32 (1H, m, H-2'), 6.34 (1H, d, $J=4.6$ Hz, H-1'), 8.00 (1H, s, H-2 or 8), 8.02 (1H, s, H-8 or 2). **8**: amorphous; $[\alpha]_{\text{D}}^{20} -37.5^\circ$ (c 0.4); 274 nm (28900); δ 2.13 (1H, ddd, H-3'), 2.41 (1H, m, H-3''), 3.33 (1H, dd, H-5'), 3.41 (1H, dd, H-5''), 3.77 (12H, s, O-CH₃), 4.53 (1H, m, H-4'), 4.82 (1H, m, H-2'), 5.98 (1H, d, $J=2.2$ Hz, H-1'), 7.94 (1H, s, H-2 or 8), 8.03 (1H, s, H-8 or 2); 2110 cm⁻¹ (N₃). **9**: 203—204 °C (dec.); $[\alpha]_{\text{D}}^{22} -66.0^\circ$ (c 0.6, DMF); 258 nm (15000); δ (DMSO-d₆) 2.14 (1H, m, H-3'), 2.46 (1H, m, H-3''), 3.54 (1H, m, H-5'), 3.72 (1H, m, H-5''), 4.32 (1H, m, H-4'), 4.85 (1H, m, H-2'), 5.25 (1H, t, OH), 6.01 (1H, d, $J=3.2$ Hz, H-1'), 7.36 (2H, br s, NH₂), 8.16 (1H, s, H-2 or 8), 8.40 (1H, s, H-8 or 2); 2120 cm⁻¹ (N₃).
- 9) R. G. Pearson, J. Am. Chem. Soc., **85**, 3533 (1963); T.-L. Ho, Tetrahedron, **41**, 3 (1985).
- 10) (a) Y. Ishido, N. Sakairi, K. Okazaki, and N. Nakazaki, J. Chem. Soc., Perkin Trans. 1, **1980**, 563. (b) M. Kawana, Chem. Lett., **1981**, 1541. (c) M. J. Robins, P. Sporns, and W. H. Muhs, Can. J. Chem., **57**, 274 (1979).

(Received in Japan 30 April 1987; accepted 18 June 1987)